

RS Components

Chemwatch: **5411-68** Version No: **2.1.1.1** Safety Data Sheet according to HSNO Regulations Chemwatch Hazard Alert Code: 3 Issue Date: 15/07/2020 Print Date: 03/09/2020 L.GHS.NZL.EN

SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier

Product name	Product name RS- 3M Scotchcast 470W Resin (Part A) #811-2760, #811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)	
Synonyms	811-2760; 811-2772; 811-2776; 811-2782; 811-2785; 811-2788	
Other means of identification	Not Available	

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Casting resin.
Relevant luentineu uses	Use according to manufacturer's directions.

Details of the supplier of the safety data sheet

RS Components	
PO Box 12-127 Penrose, Auckland New Zealand	
+64 27 4747122	
+64 9 579 1700	
www.nz.ts-online.com	
Not Available	

Emergency telephone number

Association / Organisation	CHEMWATCH EMERGENCY RESPONSE
Emergency telephone numbers	+61 2 9186 1132
Other emergency telephone numbers	+64 800 700 112

Once connected and if the message is not in your prefered language then please dial 01

SECTION 2 Hazards identification

Classification of the substance or mixture

Classification ^[1]	Skin Corrosion/Irritation Category 2, Eye Irritation Category 2	
Legend:	1. Classified by Chemwatch; 2. Classification drawn from CCID EPA NZ; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI	
Determined by Chemwatch using GHS/HSNO criteria	6.3A, 6.4A	
abel elements		
	\land	

Hazard pictogram(s)	
Signal word	Warning
Hazard statement(s)	
H315	Causes skin irritation.
H319	Causes serious eye irritation.

Precautionary statement(s) Prevention

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P280 Wear protective gloves/protective clothing/eye protection/face protection.

Precautionary statement(s) Response

	•
P321	Specific treatment (see advice on this label).
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P337+P313	If eye irritation persists: Get medical advice/attention.
P302+P352	IF ON SKIN: Wash with plenty of water and soap.

Precautionary statement(s) Storage

Not Applicable

Precautionary statement(s) Disposal

Not Applicable

SECTION 3 Composition / information on ingredients

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
8001-79-4	50-65	castor oil
25322-69-4	10-25	polypropylene glycol
25791-96-2	5-15	polypropylene glycol glyceryl ether
Not Available	1-10	polyesterpolyol
1318-02-1	1-10	zeolites
Not Available	0.1-1	grey pigment
Not Available	0.1-1	polymeric carbodiimide
64742-95-6	0.1-1	naphtha petroleum. light aromatic solvent
64742-47-8	0.1-1	distillates, petroleum, light, hydrotreated

SECTION 4 First aid measures

Description of first aid measures

Eye Contact	 If this product comes in contact with the eyes: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin contact occurs: If skin contact occurs: If skin contact occurs: If skin contact occurs: If skin and hair with running water (and soap if available). Seek medical attention in event of irritation. For thermal burns: Decontaminate area around burn. Consider the use of cold packs and topical antibiotics. For first-degree burns (affecting top layer of skin) If hold burned skin under cool (not cold) running water or immerse in cool water until pain subsides. If Use compresses if running water is not available. Cover with sterile non-adhesive bandage or clean cloth. Do NOT apply butter or ointments; this may cause infection. Give over-the counter pain relivers' fragin increases or swelling, redness, fever occur. For second-degree burns (affecting top two layers of skin) Cover with sterile non-adhesive bandage or clean cloth. Do NOT apply butter or ointments; this may cause infection. Give over-the counter pain relivers' fragin increases or swelling, redness, fever occur. For second-degree burns (affecting top two layers of skin) Cover when see in cold running water is not available. Do NOT apply butter or ointments; this may cause infection. Do NOT apply butter or ointments; this may cause infection. Do NOT apply butter or ointments; this may cause infection. Porteact burns by cover loosely with sterile, nonstick bandage and secure in place with gauze or tape. Do NOT apply loce as this may lower body temperature and cause further damage. Do NOT apply loce as the person has a head, neck, or leg injury, or it would cause discomfort): Lay the person flat. Elevate feet about 12 inches. Cover the person with coat or blanket. Cover the person with assistance. In the mean time: Protect burn area cover loosely with sterile, nonstick bandage or, f

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	Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid p Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocke Perform CPR if necessary. Transport to hospital, or doctor, without delay. 	
	Ingestion	 If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway an Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Seek medical advice. Avoid giving milk or oils. Avoid giving alcohol. 	d prevent aspiration.

Indication of any immediate medical attention and special treatment needed

Any material aspirated during vomiting may produce lung injury. Therefore emesis should not be induced mechanically or pharmacologically. Mechanical means should be used if it is considered necessary to evacuate the stomach contents; these include gastric lavage after endotracheal intubation. If spontaneous vomiting has occurred after ingestion, the patient should be monitored for difficult breathing, as adverse effects of aspiration into the lungs may be delayed up to 48 hours.

for stimulants:

Treatment and Management.

A specific antidote does not exist for acute stimulant intoxication. Activated charcoal should be prescribed in a case of acute overdose. Otherwise the treatment should target specific signs and symptoms such as hypertension, agitation, seizures, and hyperthermia. Rapid supportive treatment may reduce mortality.

Supportive therapy

Acute intoxication usually presents with increased sensitivity to sensorial stimuli and paranoia. As such, decreasing the patient's level of stimulation (keep voice low, dim lights,

minimise touch) and working with the patient's paranoid state (reduce eye contact, respect personal space, do not approach from behind) is important. As in all cases of suspected poisoning, follow the ABCDEs of emergency medicine (airway, breathing, circulation, disability, exposure), then the ABCDEs of toxicology (antidotes,

basics, change absorption, change distribution, change elimination).

Decontamination with gastric lavage may be appropriate in cases of recent ingestion.

Monitor vital signs and hydrate with intravenous fluids.

Withdrawal related insomnia may be treated with trazodone (75-200 mg), hydroxyzine (25-50 mg), or diphenhydramine (50-100 mg) at bedtime.

Benzodiazepines should be avoided unless the patient is also in detox from alcohol/benzodiazepines/opiates.

Neuroleptics may be used for the symptomatic treatment of psychosis.

Physical restraints may be required in certain cases.

Common withdrawal symptoms may include dysphoria, anxiety, and irritability, decreased energy (manifested as reported fatigue, psychomotor retardation and hypersomnia), hyperphagia, decreased concentration, and paranoia. The withdrawal symptoms are uncomfortable but not life threatening; consequently, no current recommendations for a stimulant-detoxification regimen are available.

Stimulant withdrawal dysphoria is common and does not in itself represent an indication for an antidepressant. However, a thorough assessment (including consideration of an antidepressant) is recommended for persistent (longer than a week) depressive symptoms at a level of moderate or severe or associated with suicidal ideation/attempts. Medscape

Treat symptomatically.

Polyethylene glycols are generally poorly absorbed orally and are mostly unchanged by the kidney.

- Dermal absorption can occur across damaged skin (e.g. through burns) leading to increased osmolality, anion gap metabolic acidosis, elevated calcium, low ionised calcium, CNS depression and renal failure.
- Treatment consists of supportive care.

[Ellenhorn and Barceloux: Medical Toxicology]

SECTION 5 Firefighting measures

Extinguishing media

- Alcohol stable foam.
- Dry chemical powder.
- BCF (where regulations permit).
- Carbon dioxide.

Special hazards arising from the substrate or mixture

Fire Incompatibility	Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result
Advice for firefighters	
Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Prevent, by any means available, spillage from entering drains or water course. Use water delivered as a fine spray to control fire and cool adjacent area.
Fire/Explosion Hazard	 Combustible. Slight fire hazard when exposed to heat or flame. Heating may cause expansion or decomposition leading to violent rupture of containers. On combustion, may emit toxic fumes of carbon monoxide (CO). Combustion products include: carbon dioxide (CO2) acrolein nitrogen oxides (NOX) other pyrolysis products typical of burning organic material. May emit poisonous fumes. CARE: Water in contact with hot liquid may cause foaming and a steam explosion with wide scattering of hot oil and possible severe burns. Foaming may cause overflow of containers and may result in possible fire.

SECTION 6 Accidental release measures

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See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Slippery when spilt. Remove all ignition sources. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment.
Major Spills	 Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Prevent, by all means available, spillage from entering drains or water courses. Slippery when spilt.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

Precautions for safe handling	
Safe handling	 DO NOT allow clothing wet with material to stay in contact with skin Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. Prevent concentration in hollows and sumps.
Other information	Consider storage under inert gas. Refrigerated storage normally required. Ethoxylates/ alkoxylates react slowly with air or oxygen and may generate potentially sensitising intermediates (haptens) Storage under heated conditions in the presence of air or oxygen increases reaction rate. For example, after storing at 95 F/ 35 C for 30 days in the presence of air, there is measurable oxidation of the ethoxylate. Lower temperatures will allow for longer storage time and higher temperatures will shorten the storage time if stored under an air or oxygen atmosphere. • Store in original containers. • Keep containers securely sealed. • No smoking, naked lights or ignition sources. • Store in a cool, dry, well-ventilated area.

Conditions for safe storage, including any incompatibilities

Suitable container	 For ethoxylates suitable containers include carbon steel coated with baked phenolic. Any moisture may cause rusting of carbon steel. If product is moisture free, uncoated carbon steel tanks may be used. Glass container is suitable for laboratory quantities DO NOT use aluminium or galvanised containers Polyethylene or polypropylene container. Packing as recommended by manufacturer. Check all containers are clearly labelled and free from leaks.
Storage incompatibility	 Avoid reaction with oxidising agents, bases and strong reducing agents. Avoid strong acids, acid chlorides, acid anhydrides and chloroformates.

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
New Zealand Workplace Exposure Standards (WES)	distillates, petroleum, light, hydrotreated	Oil mist, mineral	5 mg/m3	10 mg/m3	Not Available	om-Sampled by a method that does not collect vapour.

Emergency Limits

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
polypropylene glycol	Polypropylene glycols	30 mg/m3	330 mg/m3	2,000 mg/m3
zeolites	Zeolites, NaA	30 mg/m3	330 mg/m3	2,000 mg/m3
zeolites	Zeolites, NaX	30 mg/m3	330 mg/m3	2,000 mg/m3
naphtha petroleum, light aromatic solvent	Naphtha (coal tar); includes solvent naphtha, petroleum (64742-88-7), naphtha (petroleum) light aliphatic, rubber solvent (64742-89-8), heaevy catalytic cracked (64741-54-4), light straight run (64741-46-4), heavy aliphatic solvent (64742-96-7), high flash aromatic and aromatic solvent naphtha (64742-95-6)	1,200 mg/m3	6,700 mg/m3	40,000 mg/m3
distillates, petroleum, light, hydrotreated	Mineral oil, heavy or light; (paraffin oil; Deobase, deodorized; heavy paraffinic; heavy naphthenic); distillates; includes 64741-53-3, 64741-88-4, 8042-47-5, 8012-95-1; 64742-54-7	140 mg/m3	1,500 mg/m3	8,900 mg/m3

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Ingredient	Original IDLH	Revised IDLH	
castor oil	Not Available	Not Available	
polypropylene glycol	Not Available	Not Available	
polypropylene glycol glyceryl ether	Not Available	Not Available	
zeolites	Not Available	Not Available	
naphtha petroleum, light aromatic solvent	Not Available	Not Available	
distillates, petroleum, light, hydrotreated	2,500 mg/m3	Not Available	
Occupational Exposure Banding			
Ingredient	Occupational Exposure Band Rating	Occupational Exposure Band Limit	
castor oil	E	≤ 0.1 ppm	
naphtha petroleum, light aromatic solvent	E	≤ 0.1 ppm	

Occupational exposure banding is a process of assigning chemicals into specific categories or bands based on a chemical's potency and the adverse health outcomes associated with exposure. The output of this process is an occupational exposure band (OEB), which corresponds to a range of exposure concentrations that are expected to protect worker health.

MATERIAL DATA

Notes:

NOTE H: Special requirements exist in relation to classification and labelling of this substance. This note applies to certain coal- and oil -derived substances and to certain entries for groups of substances in Annex VI. European Union (EU) List of harmonised classification and labelling hazardous substances, Table 3.1, Annex VI, Regulation (EC) No 1272/2008 (CLP) - up to the latest ATP

NOTE P: The classification as a carcinogen need not apply if it can be shown that the substance contains less than 0.01% w/w benzene (EINECS No 200-753-7). Note E shall also apply when the substance is classified as a carcinogen. This note applies only to certain complex oil-derived substances in Annex VI. European Union (EU) List of harmonised classification and labelling hazardous substances, Table 3.1, Annex VI, Regulation (EC) No 1272/2008 (CLP) - up to the latest ATP

Exposure controls

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.			
Personal protection				
Eye and face protection	 No special equipment for minor exposure i.e. when handling small quantities. OTHERWISE: Safety glasses with side shields. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. 			
Skin protection	See Hand protection below			
Hands/feet protection	No special equipment needed when handling small quantities. OTHERWISE: Wear general protective gloves, e.g. light weight rubber gloves.			
Body protection	See Other protection below			
Other protection	 Overalls. P.V.C apron. Barrier cream. Skin cleansing cream. 			

Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the computergenerated selection:

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Material	CPI
NEOPRENE	А

* CPI - Chemwatch Performance Index

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

Respiratory protection

Type A-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

A: Best Selection

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	A-AUS P2	-	A-PAPR-AUS / Class 1 P2
up to 50 x ES	-	A-AUS / Class 1 P2	-
up to 100 x ES	-	A-2 P2	A-PAPR-2 P2 ^

^ - Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

- Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance	Grey liquid with a castor oil odour; does not mix with water.		
Physical state	Liquid	Relative density (Water = 1)	0.99-1.02
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Applicable	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Applicable	Viscosity (cSt)	1500-1700
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	>=200	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Applicable
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Immiscible	pH as a solution (1%)	Not Applicable
Vapour density (Air = 1)	Not Available	VOC g/L	Not Applicable

SECTION 10 Stability and reactivity

Reactivity	See section 7	
Chemical stability	Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.	
Possibility of hazardous reactions	See section 7	
Conditions to avoid	e section 7	
Incompatible materials	See section 7	
Hazardous decomposition products	See section 5	

SECTION 11 Toxicological information

Information on toxicological effects

Evidence shows, or practical experience predicts, that the material produces irritation of the respiratory system, in a substantial number of

Inhaled

	individuals, following inhalation. In contrast to most organs, the lung is at irritant and then repairing the damage. The repair process, which initially may however, produce further lung damage resulting in the impairment of irritation often results in an inflammatory response involving the recruitme system. Inhalation of vapours may cause drowsiness and dizziness. This may be coordination and vertigo. Inhalation of vapours or aerosols (mists, fumes), generated by the materi of the individual. Exposure to aliphatic alcohols with more than 3 carbons may produce ce muscle weakness, delirium, CNS depression, coma, seizure, and neurob Respiratory tract involvement may produce irritation of the mucosa, respi depression, pulmonary oedema, chemical pneumonitis and bronchitis. Ca Inhalation hazard is increased at higher temperatures.	evolved to protect mammalian lungs from foreign matter and antigens, f gas exchange, the primary function of the lungs. Respiratory tract ant and activation of many cell types, mainly derived from the vascular accompanied by narcosis, reduced alertness, loss of reflexes, lack of al during the course of normal handling, may be damaging to the health ntral nervous system effects such as headache, dizziness, drowsiness, ehavioural changes. Symptoms are more acute with higher alcohols. ratory insufficiency, respiratory depression secondary to CNS	
Ingestion	Castor oil is considered minimally toxic when administered orally to huma et al., 1976). As a purgative, castor oil is ingested as a bolus. Since this gastrointestinal tract than would occur with dietary exposure, it is not surry wet faeces. The toxic effects of glycols (dihydric alcohols), following ingestion are sim (CNS), nausea, vomiting and degenerative changes in liver and kidney. Constant use of purgatives/laxatives may decrease the sensitivity of the in redevelopment of a normal habit is thus prevented. Adverse effects associated with the administration of central nervous syst laryngospasm. At low levels persons may demonstrate elevated mood, activity, and alert predominate. Seizures may occur as well and occur more commonly and Fatty acid esters are relatively non-toxic in rats. Large doses of 20-60 gm JECFA established an acceptable daily intake (ADI) of 0-25 mg/kg bw for to 3 glycerol units and an ADI of 0-7.5 mg/kg bw for polyglyceryl esters of	would lead to higher concentrations of ricinoleic acid in the brising that in an occupational setting there is no indications of loose or illar to those of alcohol, with depression of the central nervous system intestinal mucosa causing a diminished response to normal stimulii. The tem stimulants include dyspnea, coughing, bronchospasm, and mess. However, at toxic levels, irritability, insomnia, and agitation I at lower serum levels in cases of chronic overdose. //kg are lethal in rats. polyglyceryl esters of fatty acids having an average chain length of up f interesterified ricinoleic acid.	
	In the EU, the esters are listed as food additives at concentrations between 5000 and 10,000 mg/kg in certain foods, and up to 7% free glycerol/polyglycerol is allowed (i.e., 700 mg/kg). Effects on the nervous system characterise over-exposure to higher aliphatic alcohols. These include headache, muscle weakness, giddiness ataxia, (loss of muscle coordination), confusion, delirium and coma. Gastrointestinal effects may include nausea, vomiting and diarrhoea. In the absence of effective treatment, respiratory arrest is the most common cause of death in animals acutely poisoned by the higher alcohols. Accidental ingestion of the material may be damaging to the health of the individual.		
Skin Contact	Skin contact with the material may damage the health of the individual; systemic effects may result following absorption. Daily application of 0.5 ml of castor oil to the skin of adult female albino rabbits produced mild irritant reactions, including slight erythema and edema, acanthosis and disorganization of the basal layer, and slight inflammation of the dermis (Rantuccio et al., 1981) Most liquid alcohols appear to act as primary skin irritants in humans. Significant percutaneous absorption occurs in rabbits but not apparently in man. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream through, for example, cuts, abrasions, puncture wounds or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. Evidence exists, or practical experience predicts, that the material either produces inflammation of the skin in a substantial number of individuals following direct contact, and/or produces significant inflammation when applied to the healthy intact skin of animals, for up to four hours, such inflammation being present twenty-four hours or more after the end of the exposure period. Skin irritation may also be present after prolonged or repeated exposure; this may result in a form of contact dermatitis (nonallergic). The dermatitis is often characterised by skin redness (erythema) and swelling (oedema) which may progress to blistering (vesiculation), scaling and thickening of the epidermis. At the microscopic level there may be intercellular oedema of the spongy layer of the skin (spongiosis) and intracellular oedema of the epidermis.		
Eye	Evidence exists, or practical experience predicts, that the material may cause eye irritation in a substantial number of individuals. Repeated or prolonged eye contact may cause inflammation (similar to windburn) characterised by a temporary redness of the conjunctiva (conjunctivitis); temporary impairment of vision and/or other transient eye damage/ulceration may occur.		
Chronic	 Long-term exposure to respiratory irritants may result in disease of the airways involving difficult breathing and related systemic problems. Limited evidence suggests that repeated or long-term occupational exposure may produce cumulative health effects involving organs or biochemical systems. There exists limited evidence that shows that skin contact with the material is capable either of inducing a sensitisation reaction in a significant number of individuals, and/or of producing positive response in experimental animals. Glyceryl triesters (triglycerides), following ingestion, are metabolised to monoglycerides, free fatty acids and glycerol, all of which are absorbed in the intestinal mucosa and undergo further metabolism. Medium chain triglycerides (C8-C10) appear to have relatively rapid metabolism and elimination from blood and tissues compared to long chain triglycerides (C16-C18). Little or no acute, subchronic or chronic oral toxicity was seen in animal studies unless levels approached a significant percentage of calorific intake. Subcutaneous injections of tricaprylin in rats over a five-week period caused granulomatous reaction characterised by oil deposits surrounded by macrophages. Prolonged use of purgatives/ laxatives may produce watery diarrhoea with excessive loss of water and electrolytes (particularly potassium), muscular weakness and weight loss. Changes in intestinal musculature associated with malabsorption, and dilation of the bowel, similar to ulcerative colitis and to megacolon may also result. Cardiac and renal symptoms have also been reported. On the basis, primarily, of animal experiments, the material may be regarded as carcinogenic to humans. At least one classification body 		
	considers that there is sufficient evidence to provide a strong presumptio of: - appropriate long-term animal studies - other relevant information	n that human exposure to the material may result in cancer on the basis	
RS- 3M Scotchcast 470W			
Resin (Part A) #811-2760,	ΤΟΧΙΟΙΤΥ	IRRITATION	
#811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)	Not Available	Not Available	
	TOXICITY	IRRITATION	
castor oil	Not Available	Eye (rabbit): 500 mg mild	
		Skin (human): 50 mg/48h mild	

	TOXICITY	IRRITATION
	Oral (rat) LD50: >2000 mg/kg ^[1]	Eye: no adverse effect observed (not irritating) ^[1]
polypropylene glycol	Oral (rat) LD50: >5000 mg/kg ^[1]	Skin (rabbit): 500 mg mild
		Skin: no adverse effect observed (not irritating) ^[1]
	ΤΟΧΙΟΙΤΥ	IRRITATION
	Dermal (rabbit) LD50: >20000 mg/kg ^[2]	Eye: no adverse effect observed (not irritating) ^[1]
lypropylene glycol glyceryl ether	Inhalation (rat) LC50: >200 mg/l/h* ^[2]	Eye: non-irritant *
ether	Oral (rat) LD50: 2830 mg/kg ^[2]	Skin (rabbit): 500 mg (open)-mild
		Skin: no adverse effect observed (not irritating) $\left[1 \right]$
	тохісіту	IRRITATION
	>4.575 mg/l/1hr ^[2]	Not Available
	Dermal (rabbit) LD50: >2000 mg/kg ^[2]	
zeolites	Oral (rat) LD50: >27000 mg/kg ^[2]	
	Oral (rat) LD50: >5110 mg/kg ^[2]	
	Oral (rat) LD50: 5000 mg/kg ^[2]	
	ΤΟΧΙΟΙΤΥ	IRRITATION
	Inhalation (rat) LC50: >7331.62506 mg/l/8h*[2]	Eye: no adverse effect observed (not irritating) ^[1]
	Oral (rat) LD50: >4500 mg/kg ^[1]	Skin: adverse effect observed (irritating) ^[1]
naphtha petroleum, light	Oral (rat) LD50: >5000 mg/kg ^[1]	
aromatic solvent	Oral (rat) LD50: >5570 mg/kg ^[1]	
	Oral (rat) LD50: >7000 mg/kg ^[1]	
	Oral (rat) LD50: 14063 mg/kg ^[1]	
	Oral (rat) LD50: 6620 mg/kg ^[1]	
	ΤΟΧΙΟΙΤΥ	IRRITATION
distillates, petroleum, light, hydrotreated	Not Available	Eye: no adverse effect observed (not irritating) ^[1]
nyurotreateu		Skin: adverse effect observed (irritating) ^[1]
Legend:	1. Value obtained from Europe ECHA Registered Substances specified data extracted from RTECS - Register of Toxic Effect	- Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwis at of chemical Substances
I		

The acute oral LD50 values in rats for both were greater than >2000 mg/kg bw Clinical signs were generally associated with poor condition following administration of high doses (salivation, diarrhoea, staining, piloerection and lethargy). There were no adverse effects on body weight in any study In some studies, excess test substance and/or irritation in the gastrointestinal tract was observed at necropsy. Skin and eye irritation potential, with a few stated exceptions, is chain length dependent and decreases with increasing chain length According to several OECD test regimes the animal skin irritation studies indicate that the C6-10 aliphatic acids are severely irritating or corrosive, while the C12 aliphatic acid is irritating, and the C14-22 aliphatic acids generally are not irritating or mildly irritating. Human skin irritation studies using more realistic exposures (30-minute,1-hour or 24-hours) indicate that the aliphatic acids have sufficient, good or very good skin compatibility Animal eye irritation studies indicate that among the aliphatic acids, the C8-12 aliphatic acids are irritating to the eye while the C14-22 aliphatic acids are not irritating. Eye irritation potential of the ammonium salts does not follow chain length dependence; the C18 ammonium salts are corrosive to the eyes. Dermal absorption: The in vitro penetration of C10, C12, C14, C16 and C18 fatty acids (as sodium salt solutions) through rat skin decreases with increasing chain length. At 86.73 ug C16/cm2 and 91.84 ug C18/cm2, about 0.23% and less than 0.1% of the C16 and C18 soap solutions is absorbed after 24 h exposure, respectively. CASTOR OIL Sensitisation: No sensitisation data were located. Repeat dose toxicity: Repeated dose oral (gavage or diet) exposure to aliphatic acids did not result in systemic toxicity with NOAELs greater than the limit dose of 1000 mg/kg bw For Group E aliphatic esters (polyol esters): According to a classification scheme described by the American Chemistry Council' Aliphatic Esters Panel, Group E substances are esters of monoacids, mainly common fatty acids, and trihydroxy or polyhydroxyalcohols or polyols, such as pentaerythritol (PE), 2-ethyl-2-(hydroxymethyl)-1,3-propanediol or trimethylolpropane (TMP), and dipentaerythritol (diPE). The Group E substances often are referred to as "polyol esters" The polyol esters are unique in their chemical characteristics since they lack beta-tertiary hydrogen atoms, thus leading to stability against oxidation and elimination. The fatty acids often range from C5-C10 to as high as C18 (e.g., oleic, stearic, isostearic, tall oil fatty acids) in carbon number and generally are derived from naturally occurring sources. Group E esters may have multiple ester linkages and may include mixed esters derived from different carbon-length fatty acid mixtures. For triglycerides: Carboxylic acid esters will undergo enzymatic hydrolysis by ubiquitously expressed GI esterases. The rate of hydrolysis is dependent on the structure of the ester, and may therefore be rapid or rather slow. Thus, due to hydrolysis, predictions on oral absorption based on the physicochemical characteristics of the intact parent substance alone may no longer apply. When considering the hydrolysis product glycerol, absorption is favoured based on passive and active absorption of glycerol.

	The Cosmetic Ingredient Review (CIR) Expert Panel has issued three final reports on the safe glycerin High purity is needed for the triglycerides. Previously the Panel published a final report on a of the diglyceride family are safe in the present practices of use and concentration provided the epidermal hyperplasia. The Panel discussed that there was an increased level of concern be	diglycerides, and concluded that the ingredients in content of 1,2-diesters is not high enough to induc
	kinase C (PKC) and the tumor promotion potential of 1,2-diacylglycerols. The Panel noted tha 1,2-diesters, raising the concern that 1,2-diesters could potentially induce hyperplasia. Some tumorigenic effects have been reported in animal studies using castor oil The castor seed contains ricin, a toxic protein. Heating during the oil extraction process denat harvesting castor beans may not be without risk. Allergenic compounds found on the plant su the harvest of castor beans a human health risk. The United States Food and Drug Administration (FDA) has categorized castor oil as "genera over-the-counter use as a laxative with its major site of action the small intestine where it is di	ures and inactivates the protein. However, rface can cause permanent nerve damage, makin Ily recognized as safe and effective" (GRASE) for
POLYPROPYLENE GLYCOL	** Rohm and Haas Paraplex WP-1 MSDS Polyethers, for example, ethoxylated surfactants and polyethylene glycols, are highly suscept stabilize intermediary radicals involved. Investigations of a chemically well-defined alcohol (ethoxylate, showed that polyethers form complex mixtures of oxidation products when expose Sensitization studies in guinea pigs revealed that the pure nonoxidized surfactant itself is non- oxidation products are sensitizers. Two hydroperoxides were identified in the oxidation mixture pentaoxaheptacosan-1-ol) was stable enough to be isolated. It was found to be a strong sen detection of sensitization capacity). The material may be irritating to the eye, with prolonged contact causing inflammation. Repeat conjunctivitis. The material may cause skin irritation after prolonged or repeated exposure and may produce dermatitis is often characterised by skin redness (erythema) and swelling the epidermis. Histo spongy layer (spongiosis) and intracellular oedema of the epidermis.	ible towards air oxidation as the ether oxygens wi pentaethylene glycol mono-n-dodecyl ether) ed to air. sensitizing but that many of the investigated e, but only one (16-hydroperoxy-3,6,9,12,15- sitizer in LLNA (local lymph node assay for ated or prolonged exposure to irritants may produc e a contact dermatitis (nonallergic). This form of
POLYPROPYLENE GLYCOL GLYCERYL ETHER	The material may cause skin irritation after prolonged or repeated exposure and may produce a contact dermatitis (nonallergic). This form of dermatitis is often characterised by skin redness (erythema) and swelling epidermis. Histologically there may be intercellular oedema of the spongy layer (spongiosis) and intracellular oedema of the epidermis. Data for Niax Polyol L-56 Data for Niax Polyol LG-168 * BASF Multranol 9175 SDS	
ZEOLITES	Inhalation (-) LC50: >18.3 mg/l/1hr for sodium aluminosilicate, zeolite A: Skin (rabbit): non-irrit	tating Eye (rabbit): slight [Grace]
	condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS. For trimethylbenzenes: Absorption of 1,2,4-trimethylbenzene occurs after oral, inhalation, or dermal exposure. Occupationally, inhalation and dermal exposures are the most important routes of absorption although systemic intoxication from dermal absorption is not likely to occur due to the dermal irritation caused by the chemical prompting quick removal. Following oral administration of the chemical to rats, 62.6% of the dose was recovered as urinary metabolites indicating substantial absorption . 1,2,4-Trimethylbenzenes - TMBs) Acute Toxicity Acute Toxicity Acute toxicity studies (oral, dermal and inhalation routes of exposure) have been conducted in rats using various solvent products containing predominantly mixed C9 aromatic hydrocarbons (CAS RN 64742-95-6). Inhalation LC50's range from 6,000 to 10,000 mg/m 3 for C9 aromatic naphtha and 18,000 to 24,000 mg/m3 for 1,2,4 and 1,3,5-TMB, respectively. A rat oral LD50 reported for 1,2,4-TMB is 5 grams/kg bw and a rat dermal LD50 for the C9 aromatic naphtha is >4 ml/kg bw. These data indicate that C9 aromatic solvents show that LD50/LC50 values are greater than limit doses for acute toxicity studies established under OECD test guidelines. Irritation and Sensitization	
NAPHTHA PETROLEUM, LIGHT AROMATIC SOLVENT	compound. Key criteria for the diagnosis of RADS include the absence of preceding respirato onset of persistent asthma-like symptoms within minutes to hours of a documented exposure spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholini lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diag For trimethylbenzenes: Absorption of 1,2,4-trimethylbenzene occurs after oral, inhalation, or dermal exposure. Occup most important routes of absorption although systemic intoxication from dermal absorption is by the chemical prompting quick removal. Following oral administration of the chemical to rats metabolites indicating substantial absorption . 1,2,4-Trimethylbenzene is lipophilic and may act For C9 aromatics (typically trimethylbenzenes - TMBs) Acute Toxicity Acute toxicity studies (oral, dermal and inhalation routes of exposure) have been conducted in predominantly mixed C9 aromatic hydrocarbons (CAS RN 64742-95-6). Inhalation LC50's ran naphtha and 18,000 to 24,000 mg/m3 for 1,2,4 and 1,3,5-TMB, respectively. A rat oral LD50 r dermal LD50 for the C9 aromatic naphtha is >4 ml/kg bw. These data indicate that C9 aromat than limit doses for acute toxicity studies established under OECD test guidelines. Irritation and Sensitization Several irritation studies, including skin, eye, and lung/respiratory system, have been conduct	to the irritant. A reversible airflow pattern, on e challenge testing and the lack of minimal gnosis of RADS. ationally, inhalation and dermal exposures are the not likely to occur due to the dermal irritation caus s, 62.6% of the dose was recovered as urinary ccumulate in fat and fatty tissues. In rats using various solvent products containing ge from 6,000 to 10,000 mg/m 3 for C9 aromatic eported for 1,2,4-TMB is 5 grams/kg bw and a rat ic solvents show that LD50/LC50 values are great
,	compound. Key criteria for the diagnosis of RADS include the absence of preceding respirato onset of persistent asthma-like symptoms within minutes to hours of a documented exposure spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholini lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diag For trimethylbenzenes: Absorption of 1,2,4-trimethylbenzene occurs after oral, inhalation, or dermal exposure. Occup most important routes of absorption although systemic intoxication from dermal absorption is by the chemical prompting quick removal. Following oral administration of the chemical to rats metabolites indicating substantial absorption . 1,2,4-Trimethylbenzene is lipophilic and may ac For C9 aromatics (typically trimethylbenzenes - TMBs) Acute Toxicity Acute toxicity studies (oral, dermal and inhalation routes of exposure) have been conducted in predominantly mixed C9 aromatic hydrocarbons (CAS RN 64742-95-6). Inhalation LCS0's ran naphtha and 18,000 to 24,000 mg/m3 for 1,2,4 and 1,3,5-TMB, respectively. A rat oral LD50 r dermal LD50 for the C9 aromatic naphtha is >4 ml/kg bw. These data indicate that C9 aromat than limit doses for acute toxicity studies established under OECD test guidelines. Irritation and Sensitization	to the irritant. A reversible airflow pattern, on e challenge testing and the lack of minimal mosis of RADS. ationally, inhalation and dermal exposures are the not likely to occur due to the dermal irritation caus s, 62.6% of the dose was recovered as urinary ccumulate in fat and fatty tissues. In rats using various solvent products containing ge from 6,000 to 10,000 mg/m 3 for C9 aromatic eported for 1,2,4-TMB is 5 grams/kg bw and a rat ic solvents show that LD50/LC50 values are great ted on members of the category. * [Devoe] . In gastrointestinal tract and that the absorption of . With respect to the carbon chain lengths likely to raffins. inal tract in various species. In many cases, the hydrocarbon absorption on concomitant triglyceric at a series of solubilising phases in the intestinal ite to the lipid phase of the intestinal absorptive ce r42-81-0) ranged from > 2 to >20 g/kg The derma awley rats for straight run kerosene (CAS No. nd > 5.2 mg/l, respectively. No mortalities in rats e (probably a desulfurised kerosene). Six hour
LIGHT AROMATIC SOLVENT	compound. Key criteria for the diagnosis of RADS include the absence of preceding respirato onset of persistent asthma-like symptoms within minutes to hours of a documented exposure spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholin lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diag For trimethylbenzenes: Absorption of 1,2,4-trimethylbenzene occurs after oral, inhalation, or dermal exposure. Occup most important routes of absorption although systemic intoxication from dermal absorption is by the chemical prompting quick removal. Following oral administration of the chemical to rats metabolites indicating substantial absorption . 1,2,4-Trimethylbenzene is lipophilic and may act For C9 aromatics (typically trimethylbenzenes - TMBs) Acute Toxicity Acute toxicity studies (oral, dermal and inhalation routes of exposure) have been conducted in predominantly mixed C9 aromatic hydrocarbons (CAS RN 64742-95-6). Inhalation LC50's ran naphtha and 18,000 to 24,000 mg/m3 for 1,2,4 and 1,3,5-TMB, respectively. A rat oral LD50 r dermal LD50 for the C9 aromatic naphtha is >4 ml/kg bw. These data indicate that C9 aromat than limit doses for acute toxicity studies established under OECD test guidelines. Irritation and Sensitization Several irritation studies, including skin, eye, and lung/respiratory system, have been conduct Studies indicate that normal, branched and cyclic paraffins are absorbed from the mammaliar n-paraffins is inversely proportional to the carbon chain length, with little absorption above C3G be present in mineral oil, n-paraffins may be absorbed to a greater extent that iso- or cyclo-pa The major classes of hydrocarbons have been shown to be well absorbed by the gastrointest hydrophobic hydrocarbons are ingested in association with dietary lipids. The dependence of digestion and absorption, is known as the "hydrocarbon continuum hypothesis", and asserts tt lumen, created by dietary triglycerides and their digestion produ	to the irritant. A reversible airflow pattern, on e challenge testing and the lack of minimal mosis of RADS. ationally, inhalation and dermal exposures are the not likely to occur due to the dermal irritation caus s, 62.6% of the dose was recovered as urinary ccumulate in fat and fatty tissues. In rats using various solvent products containing ge from 6,000 to 10,000 mg/m 3 for C9 aromatic eported for 1,2,4-TMB is 5 grams/kg bw and a rat ic solvents show that LD50/LC50 values are great ted on members of the category. * [Devoe] . In gastrointestinal tract and that the absorption of . With respect to the carbon chain lengths likely to raffins. inal tract in various species. In many cases, the hydrocarbon absorption on concomitant triglyceric at a series of solubilising phases in the intestinal ite to the lipid phase of the intestinal absorptive ce r42-81-0) ranged from > 2 to >20 g/kg The derma awley rats for straight run kerosene (CAS No. nd > 5.2 mg/l, respectively. No mortalities in rats e (probably a desulfurised kerosene). Six hour
LIGHT AROMATIC SOLVENT DISTILLATES, PETROLEUM, LIGHT, HYDROTREATED	compound. Key criteria for the diagnosis of RADS include the absence of preceding respirato onset of persistent asthma-like symptoms within minutes to hours of a documented exposure spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholin lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diag. For trimethylbenzenes: Absorption of 1.2,4-trimethylbenzene occurs after oral, inhalation, or dermal absorption is by the chemical prompting quick removal. Following oral administration of the chemical to rats metabolites indicating substantial absorption 1.2,4-Trimethylbenzene is lipophilic and may ac For C9 aromatics (typically trimethylbenzenes - TMBs) Acute Toxicity Acute toxicity studies (oral, dermal and inhalation routes of exposure) have been conducted in predominantly mixed C9 aromatic hydrocarbons (CAS RN 64742-95-6). Inhalation LC50's ran naphtha and 18,000 to 24,000 mg/m3 for 1,2,4 and 1,3,5-TMB, respectively. A rat oral LD50 r dermal LD50 for the C9 aromatic naphtha is >4 ml/kg bw. These data indicate that C9 aromat than limit doses for acute toxicity studies established under OECD test guidelines. Irritation and Sensitization Several irritation studies, including skin, eye, and lung/respiratory system, have been conduct Studies indicate that normal, branched and cyclic paraffins are absorbed from the mammaliar n-paraffins is inversely proportional to the carbon chain length, with little absorption above C30 be present in mineral oil, n-paraffins may be absorbed to a greater extent that iso- or cyclo-pa The major classes of hydrocarbons have been shown to be well absorbed by the gastrointest hydrophobic hydrocarbons are ingested in association with dietary lipids. The dependence of digestion and absorption, is known as the "hydrocarbon continuum hypothesis", and asserts tf lumen, created by dietary triglycerides and their digestion products, afford hydrocarbons a rou (enterocyte) membrane. For "kerosenes" Acute toxicity: Oral LD50s	to the irritant. A reversible airflow pattern, on e challenge testing and the lack of minimal mosis of RADS. ationally, inhalation and dermal exposures are the not likely to occur due to the dermal irritation caus s, 62.6% of the dose was recovered as urinary ccumulate in fat and fatty tissues. In rats using various solvent products containing ge from 6,000 to 10,000 mg/m 3 for C9 aromatic eported for 1,2,4-TMB is 5 grams/kg bw and a rat ic solvents show that LD50/LC50 values are great ted on members of the category. * [Devoe] . In gastrointestinal tract and that the absorption of . With respect to the carbon chain lengths likely to raffins. inal tract in various species. In many cases, the hydrocarbon absorption on concomitant triglyceric at a series of solubilising phases in the intestinal ite to the lipid phase of the intestinal absorptive ce r42-81-0) ranged from > 2 to >20 g/kg The derma awley rats for straight run kerosene (CAS No. nd > 5.2 mg/l, respectively. No mortalities in rats e (probably a desulfurised kerosene). Six hour
LIGHT AROMATIC SOLVENT DISTILLATES, PETROLEUM, LIGHT, HYDROTREATED CASTOR OIL & DISTILLATES, PETROLEUM, LIGHT, HYDROTREATED	compound. Key criteria for the diagnosis of RADS include the absence of preceding respirato onset of persistent asthma-like symptoms within minutes to hours of a documented exposure spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholin lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diag For trimethylbenzenes: Absorption of 1.2,4-trimethylbenzene occurs after oral, inhalation, or dermal exposure. Occup most important routes of absorption although systemic intoxication from dermal absorption is by the chemical prompting quick removal. Following oral administration of the chemical to rats metabolites indicating substantial absorption 1,2,4-Trimethylbenzene is lipophilic and may ac For C9 aromatics (typically trimethylbenzenes - TMBs) Acute Toxicity Acute Toxicity acute Toxicity studies (oral, dermal and inhalation routes of exposure) have been conducted in predominantly mixed C9 aromatic hydrocarbons (CAS RN 64742-95-6). Inhalation LC50's ran naphtha and 18,000 to 24,000 mg/m3 for 1,2,4 and 1,3,5-TMB, respectively. A rat oral LD50 r dermal LD50 for the C9 aromatic naphtha is >4 ml/kg bw. These data indicate that C9 aromat than limit doses for acute toxicity studies established under OECD test guidelines. Irritation and Sensitization Several irritation studies, including skin, eye, and lung/respiratory system, have been conduct Studies indicate that normal, branched and cyclic paraffins are absorbed from the mammaliar n-paraffins is inversely proportional to the carbon chain length, with little absorption above C33 be present in mineral oil, n-paraffins may be absorbed to a greater extent that iso- or cyclo-p3 The major classes of hydrocarbons have been shown to be well absorbed by the gastrointest hydrophobic hydrocarbons are ingested in association with dietary lipids. The dependence of digestion and absorption, is known as the "hydrocarbon continuum hypothesis", and asserts tt lumen, created by dietary triglycerides and their dig	to the irritant. A reversible airflow pattern, on e challenge testing and the lack of minimal mosis of RADS. ationally, inhalation and dermal exposures are the not likely to occur due to the dermal irritation caus s, 62.6% of the dose was recovered as urinary ccumulate in fat and fatty tissues. In rats using various solvent products containing ge from 6,000 to 10,000 mg/m 3 for C9 aromatic eported for 1,2,4-TMB is 5 grams/kg bw and a rat ic solvents show that LD50/LC50 values are great red on members of the category. * [Devoe] . In gastrointestinal tract and that the absorption of D. With respect to the carbon chain lengths likely to raffins. inal tract in various species. In many cases, the hydrocarbon absorption on concomitant triglyceric tat a series of solubilising phases in the intestinal te to the lipid phase of the intestinal absorptive ce r42-81-0) ranged from > 2 to >20 g/kg The derma awley rats for straight run kerosene (CAS No. nd > 5.2 mg/l, respectively. No mortalities in rats e (probably a desulfurised kerosene). Six hour "moderate" to "severe" irritation.
LIGHT AROMATIC SOLVENT DISTILLATES, PETROLEUM, LIGHT, HYDROTREATED CASTOR OIL & DISTILLATES, PETROLEUM, LIGHT, HYDROTREATED Acute Toxicity	compound. Key criteria for the diagnosis of RADS include the absence of preceding respirato onset of persistent asthma-like symptoms within minutes to hours of a documented exposure spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholin lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diag For trimethylbenzenes: Absorption of 1,2,4-trimethylbenzene occurs after oral, inhalation, or dermal exposure. Occup most important routes of absorption although systemic intoxication from dermal absorption is by the chemical prompting quick removal. Following oral administration of the chemical to rats metabolites indicating substantial absorption . 1,2,4-trimethylbenzene is lipophilic and may at For C9 aromatics (typically trimethylbenzenes - TMBs) Acute Toxicity Acute toxicity studies (oral, dermal and inhalation routes of exposure) have been conducted in predominantly mixed C9 aromatic hydrocarbons (CAS RN 64742-95-6). Inhalation LC50's ran naphtha and 18,000 to 24,000 mg/m3 for 1,2,4 and 1,3,5-TMB, respectively. A rat oral LD50 or dermal LD50 for the C9 aromatic naphtha is >4 ml/kg bw. These data indicate that C9 aromat than limit doses for acute toxicity studies established under OECD test guidelines. Irritation and Sensitization Several irritation studies, including skin, eye, and lung/respiratory system, have been conduct Studies indicate that normal, branched and cyclic paraffins are absorbed from the mammaliar n-paraffins is inversely proportional to the carbon chain length, with little absorption above C30 be present in mineral oil, n-paraffins may be absorbed to a greater extent that iso- or cyclo-pa The major classes of hydrocarbons have been shown to be well absorbed by the gastrointest hydrophobic hydrocarbons are ingested in association with dietary lipids. The dependence of digestion and absorption, is known as the "hydrocarbon continuum hypothesis", and asserts th lumen, created by dietary triglycerides and their digestion produ	to the irritant. A reversible airflow pattern, on e challenge testing and the lack of minimal mosis of RADS. ationally, inhalation and dermal exposures are the not likely to occur due to the dermal irritation caus s, 62.6% of the dose was recovered as urinary ccumulate in fat and fatty tissues. In rats using various solvent products containing ge from 6,000 to 10,000 mg/m 3 for C9 aromatic eported for 1,2,4-TMB is 5 grams/kg bw and a rat ic solvents show that LD50/LC50 values are great red on members of the category. * [Devoe] . In gastrointestinal tract and that the absorption of 0. With respect to the carbon chain lengths likely to raffins. Inal tract in various species. In many cases, the hydrocarbon absorption on concomitant triglyceric tat a series of solubilising phases in the intestinal te to the lipid phase of the intestinal absorptive ce r42-81-0) ranged from > 2 to >20 g/kg The derma awley rats for straight run kerosene (CAS No. nd > 5.2 mg/l, respectively. No mortalities in rats e (probably a desulfurised kerosene). Six hour "moderate" to "severe" irritation.
LIGHT AROMATIC SOLVENT DISTILLATES, PETROLEUM, LIGHT, HYDROTREATED CASTOR OIL & DISTILLATES, PETROLEUM, LIGHT, HYDROTREATED Acute Toxicity Skin Irritation/Corrosion	compound. Key criteria for the diagnosis of RADS include the absence of preceding respirato onset of persistent asthma-like symptoms within minutes to hours of a documented exposure spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholin lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diag For trimethylbenzenes: Absorption of 1,2,4-trimethylbenzene occurs after oral, inhalation, or dermal exposure. Occup most important routes of absorption although systemic intoxication from dermal absorption is by the chemical prompting quick removal. Following oral administration of the chemical to rats metabolites indicating substantial absorption. 1,2,4-trimethylbenzene is lipophilic and may act For C9 aromatics (typically trimethylbenzenes - TMBs) Acute toxicity Acute toxicity studies (oral, dermal and inhalation routes of exposure) have been conducted in predominantly mixed C9 aromatic hydrocarbons (CAS RN 64742-95-6). Inhalation LCS0's ran naphtha and 18,000 to 24,000 mg/m3 for 1,2,4 and 1,3,5-TMB, respectively. A rat oral LD50 is ran naphtha and 18,000 to z4,000 mg/m3 for 1,2,4 and 1,3,5-TMB, respectively. A rat oral LD50. Irritation and Sensitization Several irritation studies, including skin, eye, and lung/respiratory system, have been conduct Studies indicate that normal, branched and cyclic paraffins are absorbed from the mammaliar n-paraffins is inversely proportional to the carbon chain length, with little absorption above C30 be present in mineral oil, n-paraffins may be absorbed to a greater extent that iso- or cyclo-pa The major classes of hydrocarbons are ingested in association with dietary lipids. The dependence of digestion and absorption, is known as the "hydrocarbon continuum hypothesis", and asserts th lumen, created by dietary triglycerides and their digestion products, afford hydrocarbons a ro (enterocyte) membrane. For "kerosenes" Acute toxicity: Oral LD50s for three kerosenes (Jet A, CAS No. 8008-20-6 and CAS No. 647 LD5	to the irritant. A reversible airflow pattern, on e challenge testing and the lack of minimal mosis of RADS. ationally, inhalation and dermal exposures are the not likely to occur due to the dermal irritation caus s, 62.6% of the dose was recovered as urinary ccumulate in fat and fatty tissues. In rats using various solvent products containing ge from 6,000 to 10,000 mg/m 3 for C9 aromatic eported for 1,2,4-TMB is 5 grams/kg bw and a rat ic solvents show that LD50/LC50 values are great ted on members of the category. * [Devoe] . In gastrointestinal tract and that the absorption of . With respect to the carbon chain lengths likely to raffins. inal tract in various species. In many cases, the hydrocarbon absorption on concomitant triglyceric tat a series of solubilising phases in the intestinal ite to the lipid phase of the intestinal absorptive ce r42-81-0) ranged from > 2 to >20 g/kg The derma awley rats for straight run kerosene (CAS No. nd > 5.2 mg/l, respectively. No mortalities in rats e (probably a desulfurised kerosene). Six hour "moderate" to "severe" irritation.

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RS- 3M Scotchcast 470W Resin (Part A) #811-2760, #811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)

Toxicity

RS- 3M Scotchcast 470W Resin (Part A) #811-2760, #811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)	Endpoint	Test Duration (hr)	Species	Value	Source
	Not Available	Not Available	Not Available	Not Available	Not Availabl
	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50	48	Crustacea	100mg/L	2
castor oil	EC50	72	Algae or other aquatic plants	>100mg/L	2
	NOEC	72	Algae or other aquatic plants	100mg/L	2
	Endpoint	Test Duration (hr)	Species	Value	Source
	LC50	96	Fish	>1-mg/L	2
polypropylene glycol	EC50	48	Crustacea	>100mg/L	2
	EC50	72	Algae or other aquatic plants	>1-mg/L	2
	NOEC	72	Algae or other aquatic plants	>=1-mg/L	2
	Endpoint	Test Duration (hr)	Species	Value	Sourc
	LC50	96	Fish	>1-mg/L	2
olypropylene glycol glyceryl	EC50	48	Crustacea	>100mg/L	2
ether	EC50	72	Algae or other aquatic plants	>100mg/L	2
	EC0	72	Algae or other aquatic plants	>=100mg/L	2
	NOEC	504	Crustacea	>=10mg/L	2
	Endpoint	Test Duration (hr)	Species	Value	Sourc
	LC50	96	Fish	1000mg/L	1
	EC50	48	Crustacea	100-1800mg/L	1
zeolites	EC50	96	Algae or other aquatic plants	18mg/L	1
	EC10	96	Algae or other aquatic plants	4.9mg/L	1
	NOEC	432	Algae or other aquatic plants	1mg/L	1
	Endpoint	Test Duration (hr)	Species	Value	Sourc
	LC50	96	Fish	4.1mg/L	2
naphtha petroleum, light aromatic solvent	EC50	48	Crustacea	3.2mg/L	2
aromatic solvent	EC50	72	Algae or other aquatic plants	>1-mg/L	2
	NOEL	72	Algae or other aquatic plants	0.1mg/L	2
	Endpoint	Test Duration (hr)	Species	Value	Sourc
	LC50	96	Fish	>1-mg/L	2
distillates, petroleum, light, hydrotreated	EC50	48	Crustacea	>1-mg/L	2
nyurotreateu	EC50	72	Algae or other aquatic plants	>1-mg/L	2
	NOEL	96	Algae or other aquatic plants	0.2mg/L	2
Legend:			ECHA Registered Substances - Ecotoxicological Informa		

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
	No Data available for all ingredients	No Data available for all ingredients

Bioaccumulative potential

Ingredient	Bioaccumulation
polypropylene glycol glyceryl ether	LOW (BCF = 7)
distillates, petroleum, light, hydrotreated	LOW (BCF = 159)

Mobility in soil

Ingredient	Mobility
	No Data available for all ingredients

SECTION 13 Disposal considerations

aste treatment methods Product / Packaging disposal	Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in the area. In some areas, certain wastes must be tracked. A Hierarchy of Controls seems to be common - the user should investigate: • Reduction • Reuse • Recycling • Disposal (if all else fails) This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. • DO NOT allow wash water from cleaning or process equipment to enter drains. • It may be necessary to collect all wash water for treatment before disposal. • In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first. • Where in doubt contact the responsible authority. • Recycle wherever possible or consult manufacturer for recycling options. • Consult State Land Waste Authority for disposal. • Bury or incinerate residue at an approved site. • Recycle containers if possible, or dispose of in an authorised landfill.
--	--

Ensure that the hazardous substance is disposed in accordance with the Hazardous Substances (Disposal) Notice 2017

Disposal Requirements

Packages that have been in direct contact with the hazardous substance must be only disposed if the hazardous substance was appropriately removed and cleaned out from the package. The package must be disposed according to the manufacturer's directions taking into account the material it is made of. Packages which hazardous content have been appropriately treated and removed may be recycled.

The hazardous substance must only be disposed if it has been treated by a method that changed the characteristics or composition of the substance and it is no longer hazardous.

SECTION 14 Transport information

Labels Required	
Marine Pollutant	NO
HAZCHEM	Not Applicable

Land transport (UN): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

This substance is to be managed using the conditions specified in an applicable Group Standard

HSR Number	Group Standard	Group Standard	
HSR002670	Surface Coatings and Colourants (Subsidiary Hazard) Group Standard 2017		
castor oil is found on the	following regulatory lists		
New Zealand Inventory of C	Chemicals (NZIoC)		
polypropylene glycol is for	ound on the following regulatory lists		
New Zealand Approved Hazardous Substances with controls New Zealand Hazardous Substances and New Organisms (HSNO) Act - Classification of Chemicals		New Zealand Hazardous Substances and New Organisms (HSNO) Act - Classification of Chemicals - Classification Data New Zealand Inventory of Chemicals (NZIoC)	
polypropylene glycol gly	ceryl ether is found on the following regulatory lists		
New Zealand Approved Hazardous Substances with controls New Zealand Hazardous Substances and New Organisms (HSNO) Act - Classification		New Zealand Hazardous Substances and New Organisms (HSNO) Act - Classification of Chemicals - Classification Data	
of Chemicals		New Zealand Inventory of Chemicals (NZIoC)	
zeolites is found on the fe	ollowing regulatory lists		
International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs		New Zealand Inventory of Chemicals (NZIoC)	
naphtha petroleum, light	aromatic solvent is found on the following regulatory lists		
Chemical Footprint Project - Chemicals of High Concern List International Agency for Research on Cancer (IARC) - Agents Classified by the IARC		New Zealand Hazardous Substances and New Organisms (HSNO) Act - Classification of Chemicals	
Monographs New Zealand Approved Hazardous Substances with controls		New Zealand Inventory of Chemicals (NZIoC)	
distillates, petroleum, ligi	ht, hydrotreated is found on the following regulatory lists		

New Zealand Hazardous Substances and New Organisms (HSNO) Act - Classification

RS- 3M Scotchcast 470W Resin (Part A) #811-2760, #811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)

of Chemicals

New Zealand Inventory of Chemicals (NZIoC) New Zealand Workplace Exposure Standards (WES)

Chemical Footprint Project - Chemicals of High Concern List	
International Agency for Research on Cancer (IARC) - Agents Classified by the IAR	С
Monographs	
International Agency for Research on Cancer (IARC) - Agents Classified by the IAR	С
Managemente Oracin du Oracina englis ta humana	

Monographs - Group 1 : Carcinogenic to humans New Zealand Approved Hazardous Substances with controls

Hazardous Substance Location

Subject to the Health and Safety at Work (Hazardous Substances) Regulations 2017.

Hazard Class	Quantity (Closed Containers)	Quantity (Open Containers)
Not Applicable	Not Applicable	Not Applicable

Certified Handler

Subject to Part 4 of the Health and Safety at Work (Hazardous Substances) Regulations 2017.

Class of substance	Quantities
Not Applicable	Not Applicable

Refer Group Standards for further information

Tracking Requirements

Not Applicable

National Inventory Status

National Inventory	Status
Australia - AIIC	Yes
Australia Non-Industrial Use	No (castor oil; polypropylene glycol; polypropylene glycol glyceryl ether; zeolites; naphtha petroleum, light aromatic solvent; distillates, petroleum, light, hydrotreated)
Canada - DSL	Yes
Canada - NDSL	No (castor oil; polypropylene glycol; polypropylene glycol glyceryl ether; naphtha petroleum, light aromatic solvent; distillates, petroleum, light, hydrotreated)
China - IECSC	Yes
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	No (zeolites)
Korea - KECI	Yes
New Zealand - NZIoC	Yes
Philippines - PICCS	Yes
USA - TSCA	Yes
Taiwan - TCSI	Yes
Mexico - INSQ	No (polypropylene glycol glyceryl ether)
Vietnam - NCI	Yes
Russia - ARIPS	Yes
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

SECTION 16 Other information

Revision Date	15/07/2020
Initial Date	15/07/2020

SDS Version Summary

Version	Issue Date	Sections Updated
2.1.1.1	15/07/2020	Appearance, Physical Properties, Supplier Information

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

- IARC: International Agency for Research on Cancer
- ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations OSF: Odour Safety Factor

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RS- 3M Scotchcast 470W Resin (Part A) #811-2760, #811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)

NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

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RS Components

Chemwatch: **5411-69** Version No: **2.1.1.1** Safety Data Sheet according to HSNO Regulations Chemwatch Hazard Alert Code: 2 Issue Date: 15/07/2020

Print Date: 03/09/2020

L.GHS.NZL.EN

SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier

Product name	RS- 3m Scotchcast 470W Resin (Part B) #811-2760, #811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)
Chemical Name	polymeric diphenylmethane diisocyanate
Synonyms	811-2760; 811-2772; 811-2776; 811-2782; 811-2785; 811-2788
Other means of identification	Not Available

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Electrical.
Relevant identified uses	Use according to manufacturer's directions.

Details of the supplier of the safety data sheet

Registered company name	RS Components
Address	PO Box 12-127 Penrose, Auckland New Zealand
Telephone	+64 27 4747122
Fax	+64 9 579 1700
Website	www.nz.rs-online.com
Email	Not Available

Emergency telephone number

Association / Organisation	CHEMWATCH EMERGENCY RESPONSE
Emergency telephone numbers	+61 2 9186 1132
Other emergency telephone numbers	+64 800 700 112

Once connected and if the message is not in your prefered language then please dial 01

SECTION 2 Hazards identification

Classification of the substance or mixture

Classification ^[1]	Acute Toxicity (Inhalation) Category 2, Skin Corrosion/Irritation Category 3, Eye Irritation Category 2, Specific target organ toxicity - single exposure Category 1, Specific target organ toxicity - repeated exposure Category 1
Legend:	1. Classified by Chemwatch; 2. Classification drawn from CCID EPA NZ; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI
Determined by Chemwatch using GHS/HSNO criteria	6.1B (inhalation), 6.3B, 6.4A, 6.9A

Label elements

Hazard pictogram(s)	

Signal word

rd Danger

Hazard statement(s)

H330	Fatal if inhaled.
H316	Causes mild skin irritation.

H319	Causes serious eye irritation.
H370	Causes damage to organs.
H372	Causes damage to organs through prolonged or repeated exposure.

Precautionary statement(s) Prevention

P260	Do not breathe mist/vapours/spray.
P271	Use only outdoors or in a well-ventilated area.
P270	Do not eat, drink or smoke when using this product.
P280	Wear protective gloves/protective clothing/eye protection/face protection.

Precautionary statement(s) Response

P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.
P308+P311	IF exposed or concerned: Call a POISON CENTER/doctor/physician/first aider.
P310	Immediately call a POISON CENTER/doctor/physician/first aider.
P320	Specific treatment is urgent (see advice on this label).

Precautionary statement(s) Storage

P403+P233	Store in a well-ventilated place. Keep container tightly closed.
P405	Store locked up.

Precautionary statement(s) Disposal

P501

Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 Composition / information on ingredients

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
9016-87-9	>=99	polymeric diphenylmethane diisocyanate

SECTION 4 First aid measures

escription of first aid measur	es
Eye Contact	 If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 If skin or hair contact occurs: Immediately flush body and clothes with large amounts of water, using safety shower if available. Quickly remove all contaminated clothing, including footwear. Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre. Transport to hospital, or doctor.
Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay. Following uptake by inhalation, move person to an area free from risk of further exposure. Oxygen or artificial respiration should be administere as needed. Asthmatic-type symptoms may develop and may be immediate or delayed up to several hours. Treatment is essentially symptomatia A physician should be consulted.
Ingestion	 IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY. For advice, contact a Poisons Information Centre or a doctor. Urgent hospital treatment is likely to be needed. In the mean time, qualified first-aid personnel should treat the patient following observation and employing supportive measures as indicate by the patient's condition. If the services of a medical officer or medical doctor are readily available, the patient should be placed in his/her care and a copy of the SD should be provided. Further action will be the responsibility of the medical specialist. If medical attention is not available on the worksite or surroundings send the patient to a hospital together with a copy of the SDS. Where medical attention is not immediately available or where the patient is more than 15 minutes from a hospital or unless instructer otherwise: INDUCE vomiting with fingers down the back of the throat, ONLY IF CONSCIOUS. Lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. NOTE: Wear a protective glove when inducing vomiting by mechanical means.

Indication of any immediate medical attention and special treatment needed

For sub-chronic and chronic exposures to isocyanates:

- This material may be a potent pulmonary sensitiser which causes bronchospasm even in patients without prior airway hyperreactivity.
- Clinical symptoms of exposure involve mucosal irritation of respiratory and gastrointestinal tracts
- Conjunctival irritation, skin inflammation (erythema, pain vesiculation) and gastrointestinal disturbances occur soon after exposure.
- Pulmonary symptoms include cough, burning, substernal pain and dyspnoea.
- Some cross-sensitivity occurs between different isocyanates.
- Noncardiogenic pulmonary oedema and bronchospasm are the most serious consequences of exposure. Markedly symptomatic patients should receive oxygen, ventilatory support and an intravenous line.
- Treatment for asthma includes inhaled sympathomimetics (epinephrine [adrenalin], terbutaline) and steroids.
- Activated charcoal (1 g/kg) and a cathartic (sorbitol, magnesium citrate) may be useful for ingestion.
- Mydriatics, systemic analgesics and topical antibiotics (Sulamyd) may be used for corneal abrasions.
- There is no effective therapy for sensitised workers.

[Ellenhorn and Barceloux; Medical Toxicology]

NOTE: Isocyanates cause airway restriction in naive individuals with the degree of response dependant on the concentration and duration of exposure. They induce smooth muscle contraction which leads to bronchoconstrictive episodes. Acute changes in lung function, such as decreased FEV1, may not represent sensitivity.

[Karol & Jin, Frontiers in Molecular Toxicology, pp 56-61, 1992]

Personnel who work with isocyanates, isocyanate prepolymers or polyisocyanates should have a pre-placement medical examination and periodic examinations thereafter, including a pulmonary function test. Anyone with a medical history of chronic respiratory disease, asthmatic or bronchial attacks, indications of allergic responses, recurrent eczema or sensitisation conditions of the skin should not handle or work with isocyanates. Anyone who develops chronic respiratory distress when working with isocyanates should be removed from exposure and examined by a physician. Further exposure must be avoided if a sensitivity to isocyanates or polyisocyanates has developed.

SECTION 5 Firefighting measures

Extinguishing media

- Small quantities of water in contact with hot liquid may react violently with generation of a large volume of rapidly expanding hot sticky semi-solid foam.
- Presents additional hazard when fire fighting in a confined space.
- Cooling with flooding quantities of water reduces this risk.
- Water spray or fog may cause frothing and should be used in large quantities.
- Foam.
- Dry chemical powder.
- BCF (where regulations permit).
- Carbon dioxide.

Special hazards arising from the substrate or mixture

Fire Incompatibility + Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

Advice for firefighters

Autor for monghere	
Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Prevent, by any means available, spillage from entering drains or water course. Use water delivered as a fine spray to control fire and cool adjacent area.
Fire/Explosion Hazard	 -Combustible. -Moderate fire hazard when exposed to heat or flame. -When heated to high temperatures decomposes rapidly generating vapour which pressures and may then rupture containers with release of flammable and highly toxic isocyanate vapour. -Burns with acrid black smoke and poisonous fumes. Combustion products include: carbon dioxide (CO2) isocyanates and minor amounts of hydrogen cyanide nitrogen oxides (NOX) other pyrolysis products typical of burning organic material. May emit corrosive fumes. When heated at high temperatures many isocyanates decompose rapidly generating a vapour which pressurises containers, possibly to the point of rupture. Release of toxic and/or flammable isocyanate vapours may then occur

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Remove all ignition sources. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment.
Major Spills	 Liquid Isocyanates and high isocyanate vapour concentrations will penetrate seals on self contained breathing apparatus - SCBA should be used inside encapsulating suit where this exposure may occur. For isocyanate spills of less than 40 litres (2 m2): Evacuate area from everybody not dealing with the emergency, keep them upwind and prevent further access, remove ignition sources and, if inside building, ventilate area as well as possible. Notify supervision and others as necessary. Put on personal protective equipment (suitable respiratory protection, face and eye protection, protective suit, gloves and impermeable boots). Control source of leakage (where applicable).

- Avoid contamination with water, alkalies and detergent solutions.
- Material reacts with water and generates gas, pressurises containers with even drum rupture resulting.
- **DO NOT** reseal container if contamination is suspected.
- Open all containers with care.
 DO NOT touch the spill material
- Moderate hazard.
- Clear area of personnel and move upwind.
- Alert Fire Brigade and tell them location and nature of hazard.
- Wear breathing apparatus plus protective gloves.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

Precautions for safe handling	
Safe handling	 DO NOT allow clothing wet with material to stay in contact with skin Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. Prevent concentration in hollows and sumps.
Other information	Consider storage under inert gas. for commercial quantities of isocyanates: -Isocyanates should be stored in adequately bunded areas. Nothing else should be kept within the same bunding. Pre-polymers need not be segregated. Drums of isocyanates should be stored under cover, out of direct sunlight, protected from rain, protected from physical damage and well away from moisture, acids and alkalis. Rotate all stock to prevent ageing. Use on FIFO (First In-First Out) basis • Store in original containers. • Keep containers securely sealed. • No smoking, naked lights or ignition sources. • Store in a cool, dry, well-ventilated area.

Conditions for safe storage, including any incompatibilities

Suitable container	 Metal can or drum Packaging as recommended by manufacturer. Check all containers are clearly labelled and free from leaks. 	
Storage incompatibility	 Arnines, water, metals. Avoid reaction with oxidising agents, bases and strong reducing agents. Avoid strong acids, acid chlorides, acid anhydrides and chloroformates. 	

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
New Zealand Workplace Exposure Standards (WES)	polymeric diphenylmethane diisocyanate	lsocyanates, all, (as -NCO)	0.02 mg/m3	0.07 mg/m3	Not Available	dsen-Dermal sensitiser (rsen)-Respiratory sensitiser Note: These values apply to all isocyanates, including prepolymers, present in the workplace air as vapours, mist or dust.

Emergency Limits						
Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3		
polymeric diphenylmethane diisocyanate	Polymethylene polyphenyl isocyanate; (Polymeric diphenylmethane diisocyanate)		0.15 mg/m3	3.6 mg/m3	22 mg/m3	
Ingredient	ngredient Original IDLH Revised IDLH					
polymeric diphenylmethane diisocyanate	Not Available	Not Available				

MATERIAL DATA

Exposure controls

Exposure controlo	
Appropriate engineering controls	 All processes in which isocyanates are used should be enclosed wherever possible. Total enclosure, accompanied by good general ventilation, should be used to keep atmospheric concentrations below the relevant exposure standards. If total enclosure of the process is not feasible, local exhaust ventilation may be necessary. Local exhaust ventilation is essential where lower molecular weight isocyanates (such as TDI or HDI) is used or where isocyanate or polyurethane is sprayed. Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.

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RS- 3m Scotchcast 470W Resin (Part B) #811-2760, #811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)

Personal protection	
Eye and face protection	 Safety glasses with side shields. Chemical goggles. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task.
Skin protection	See Hand protection below
Hands/feet protection	 NOTE: The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact. Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed. The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application. The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice. Personal hygiene is a key element of effective hand care. Do NOT wear natural rubber (latex gloves). Isocyanate resistant materials include Teffon, Viton, nitrile rubber and some PVA gloves. Protective gloves and overalls should be worn as specified in the appropriate national standard. Contaminated garments should be removed promptly and should not be re-used until they have been decontaminated. NOTE: Natural rubber, neoprene, PVC can be affected by isocyanates DO NOT use skin cream unless necessary and then use only minimum amount. Isocyanate vapour may be absorbed into skin cream and this increases hazard. Avoid contact with moisture.
Body protection	See Other protection below
Other protection	All employees working with isocyanates must be informed of the hazards from exposure to the contaminant and the precautions necessary to prevent damage to their health. They should be made aware of the need to carry out their work so that as little contamination as possible is produced, and of the importance of the proper use of all safeguards against exposure to themselves and their fellow workers. Adequate training, both in the proper execution of the task and in the use of all associated engineering controls, as well as of any personal protective equipment, is essential. Employees exposed to contamination hazards should be educated in the need for, and proper use of, facilities, clothing and equipment and thereby maintain a high standard of personal cleanliness. • Overalls. • P.V.C apron. • Barrier cream. • Skin cleansing cream.

Respiratory protection

Full face respirator with supplied air.

- Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used
- For spraying or operations which might generate aerosols:

Full face respirator with supplied air.

- In certain circumstances, personal protection of the individual employee is necessary. Personal protective devices should be regarded as being supplementary to substitution and engineering control and should not be used in preference to them as they do nothing to eliminate the hazard.
- However, in some situations, minimising exposure to isocyanates by enclosure and ventilation is not possible, and occupational exposure standards may be exceeded, particularly during on-site mixing of paints, spray-painting, foaming and maintenance of machine and ventilation systems. In these situations, air-line respirators or self-contained breathing apparatus complying with the appropriate nationals standard must be used.
- Organic vapour respirators with particulate pre- filters and powered, air-purifying respirators are NOT suitable.
- Personal protective equipment must be appropriately selected, individually fitted and workers trained in their correct use and maintenance. Personal protective equipment must be regularly checked and maintained to ensure that the worker is being protected.
- Air- line respirators or self-contained breathing apparatus complying with the appropriate national standard should be used during the clean-up of spills and the repair or clean-up of contaminated equipment and similar situations which cause emergency exposures to hazardous atmospheric concentrations of isocyanate.

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance	Brown liquid with an earthy musty odour; does not mix with water.			
Physical state	Liquid	Relative density (Water = 1)	1.2-1.24	
Odour	Not Available	Partition coefficient n-octanol / water	Not Available	
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available	
pH (as supplied)	Not Applicable	Decomposition temperature	Not Available	
Melting point / freezing point (°C)	Not Applicable	Viscosity (cSt)	120-150	
Initial boiling point and boiling range (°C)	Not Applicable	Molecular weight (g/mol)	Not Applicable	
Flash point (°C)	Not Available	Taste	Not Available	

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RS- 3m Scotchcast 470W Resin (Part B) #811-2760, #811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)

Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Immiscible	pH as a solution (1%)	Not Applicable
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 Stability and reactivity

See section 7
 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur. Presence of elevated temperatures.
See section 7
See section 7
See section 7
See section 5

SECTION 11 Toxicological information

Information on toxicological effects

Inhaled	Strong evidence exists that exposure to the material may produce very serious irreversible damage (other than carcinogenesis, mutagenesis and teratogenesis) following a single exposure by inhalation. Evidence shows, or practical experience predicts, that the material produces irritation of the respiratory system, in a substantial number of individuals, following inhalation. In contrast to most organs, the lung is able to respond to a chemical insult by first removing or neutralising the irritant and then repairing the damage. The repair process, which initially evolved to protect mammalian lungs from foreign matter and antigens, may however, produce further lung damage resulting in the impairment of gas exchange, the primary function of the lungs. Respiratory tract irritation often results in an inflammatory response involving the recruitment and activation of many cell types, mainly derived from the vascular system. The vapour/mist may be highly irritating to the upper respiratory tract and lungs; the response may be severe enough to produce bronchitis and pulmonary oedema. Possible neurological symptoms arising from isocyanate exposure include headache, insomnia, euphoria, ataxia, anxiety neurosis, depression and paranoia. Gastrointestinal disturbances are characterised by nausea and vomiting. Pulmonary sensitisation may produce asthmatic reactions ranging from inor breathing difficulties to severe allergic attacks; this may occur following a single acute exposure or may develop without warning for several hours after exposure. Inhalation of aerosols (mists, fumes), generated by the material during the course of normal handling, may produce severely toxic effects. Relatively small amounts absorbed from the lungs may prove fatal.
Ingestion	Strong evidence exists that exposure to the material may produce very serious irreversible damage (other than carcinogenesis, mutagenesis and teratogenesis) following a single exposure by swallowing. Accidental ingestion of the material may be seriously damaging to the health of the individual; animal experiments indicate that ingestion of less than 40 gram may be fatal.
Skin Contact	Evidence exists, or practical experience predicts, that the material either produces inflammation of the skin in a substantial number of individuals following direct contact, and/or produces significant inflammation when applied to the healthy intact skin of animals, for up to four hours, such inflammation being present twenty-four hours or more after the end of the exposure period. Skin irritation may also be present after prolonged or repeated exposure; this may result in a form of contact dermatitis (nonallergic). The dermatitis is often characterised by skin redness (erythema) and swelling (oedema) which may progress to blistering (vesiculation), scaling and thickening of the epidermis. At the microscopic level there may be intercellular oedema of the spongy layer of the skin (spongiosis) and intracellular oedema of the epidermis. Strong evidence exists that exposure to the material may produce very serious irreversible damage (other than carcinogenesis, mutagenesis and teratogenesis) following a single exposure by skin contact. The material may accentuate any pre-existing dermatitis condition Skin contact with the material may damage the health of the individual; systemic effects may result following absorption. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream through, for example, cuts, abrasions, puncture wounds or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material ensure that any external damage is suitably protected.
Eye	Evidence exists, or practical experience predicts, that the material may cause eye irritation in a substantial number of individuals and/or may produce significant ocular lesions which are present twenty-four hours or more after instillation into the eye(s) of experimental animals. Repeated or prolonged eye contact may cause inflammation characterised by a temporary redness (similar to windburn) of the conjunctiva (conjunctivitis); temporary impairment of vision and/or other transient eye damage/ulceration may occur.
Chronic	On the basis, primarily, of animal experiments, concern has been expressed that the material may produce carcinogenic or mutagenic effects; in respect of the available information, however, there presently exists inadequate data for making a satisfactory assessment. Long-term exposure to respiratory irritants may result in disease of the airways involving difficult breathing and related systemic problems. Practical evidence shows that inhalation of the material is capable of inducing a sensitisation reaction in a substantial number of individuals at a greater frequency than would be expected from the response of a normal population. Pulmonary sensitisation, resulting in hyperactive airway dysfunction and pulmonary allergy may be accompanied by fatigue, malaise and aching. Significant symptoms of exposure may persist for extended periods, even after exposure ceases. Symptoms can be activated by a variety of nonspecific environmental stimuli such as automobile exhaust, perfumes and passive smoking. Practical experience shows that skin contact with the material is capable either of inducing a sensitisation reaction in a substantial number of individuals, and/or of producing a positive response in experimental animals. Toxic: danger of serious damage to health by prolonged exposure through inhalation, in contact with skin and if swallowed. Serious damage (clear functional disturbance or morphological change which may have toxicological significance) is likely to be caused by
	Continued

sion No: 2.1.1.1 RS- 3m	Scotchcast 470W Resin (Part B) #811-2760, #811-2788	#811-2772, #811-2776, #811-2782, #811-2785, Print Date: 03/09/2 (NZ)
	 become apparent following direct application in subchron tests. Limited evidence suggests that repeated or long-term occ biochemical systems. Polyisocyanates still contain small amounts of monomerin monomer - have toxicological importance. In addition, sol Due to the higher molecular weight and the much lower v compared to the corresponding monomers. Nevertheless Persons with a history of asthma or other respiratory prothandling of isocyanates. The chemistry of reaction of isocyanates, as evidenced b doses to the mouth, reactions will commence at once with tract prior to reaching the stomach. Reaction products will proteins and cell components. This is corroborated by the results from an MDI inhalation A 90-day inhalation study in rats with polymeric MDI (6 hd in the nasal cavities and lungs at levels of 8 mg/m3 or gre Rats exposed for two years to a respirable aerosol of poly highest level (6 mg/m3), was there a significant incidence (adenocarcinoma). There were no lung tumours at 1 mg/m malignant and the number of animals with the tumours with prolonged respiratory irritation and the concurrent ac absence of prolonged exposure to high concentrations le occur. Isocyanate vapours/mists are irritating to the upper respiratory incoming the monomer is a significant loss from isocyanate exposure include headache, insomnia, end the concurrent incoming the stomate include headache, insomnia, end the concurrent include headache, insomnia, end the concurrent	olems or are known to be sensitised, should not be engaged in any work involving the y MDI, in biological milieu is such that in the event of a true exposure of small MDI biological macromolecules in the buccal region and will continue along the digestive I be a variety of polyureas and macromolecular conjugates with for example mucus, a study.
	difficulties to severe allergic attacks; this may occur follow	ving a single acute exposure or may develop without warning after a period of tolerance.
RS- 3m Scotchcast 470W Resin (Part B) #811-2760,		
#811-2772, #811-2776,	TOXICITY	IRRITATION
#811-2782, #811-2785, #811-2788 (NZ)	Not Available	Not Available
	ΤΟΧΙCITY	IRRITATION
polymeric diphenylmethane	Dermal (rabbit) LD50: >9400 mg/kg ^[2]	Eye (rabbit): 100 mg - mild
diisocyanate	Inhalation (rat) LC50: 0.49 mg/l/4h ^[2]	
	Oral (rat) LD50: 43000 mg/kg ^[2]	
Legend:	Value obtained from Europe ECHA Registered Substa specified data extracted from RTECS - Register of Toxic	nces - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise

1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances

POLYMERIC DIPHENYLMETHANE DIISOCYANATE	product The following information refers to contact allergens as a group and may not be specific to this product. Contact allergies quickly manifest themselves as contact eczema, more rarely as urticaria or Quincke's oedema. The pathogenesis of contact eczema involves a cell-mediated (T lymphocytes) immune reaction of the delayed type. Other allergic skin reactions, e.g. contact urticaria, involve antibody-mediated mimmune reactions. The significance of the contact allergen is not simply determined by its sensitisation potential: the distribution of the substance and the opportunities for contact with it are equally important. Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive ainways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosi of RADS. Allergic reactions which develop in the respiratory passages as bronchial asthma or thinoconjunctivitis, are mostly the result of reactions of the allergen-specific potential for causing respiratory sensitisation, the amount of the allergen, the exposure period and the genetically determined disposition of the exposed person are likely to be decisive. Factors which increase the sensitivity of the mucosa may play arole in predisposing a person to allergy. They may be genetically determined or acquired, for example, during infections or exposure to infrant substances. Particular attentition i
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RS- 3m Scotchcast 470W Resin (Part B) #811-2760, #811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)

Acute Toxicity	✓	Carcinogenicity	×
Skin Irritation/Corrosion	×	Reproductivity	×
Serious Eye Damage/Irritation	✓	STOT - Single Exposure	✓
Respiratory or Skin sensitisation	×	STOT - Repeated Exposure	*
Mutagenicity	×	Aspiration Hazard	×
		Legend: 🗙 – Data either r	not available or does not fill the criteria for classification

Data available to make classification

SECTION 12 Ecological information

Toxicity

RS- 3m Scotchcast 470W Resin (Part B) #811-2760,	Endpoint	Test Duration (hr)	Species	Value	Source
#811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)	Not Available	Not Available	Not Available	Not Available	Not Available
	Endpoint	Test Duration (hr)	Species	Value	Source
polymeric diphenylmethane	LC50	96	Fish	>1-mg/L	2
diisocyanate	EC50	72	Algae or other aquatic plants	>1-640mg/L	2
	NOEL	72	Algae or other aquatic plants	1-640mg/L	2
Legend:	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3. EPIWIN Suite V3.12 (QSAR) - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data				

for polyisocyanates:

Polyisocyanates are not readily biodegradable. However, due to other elimination mechanisms (hydrolysis, adsorption), long retention times in water are not to be expected. The resulting polyurea is more or less inert and, due to its molecular size, not bioavailable. Within the limits of water solubility, polyisocyanates have a low to moderate toxicity for aquatic organisms.

Hydrolysis would represents the primary fate mechanism for the majority of the commercial isocyanate monomers, but, is tempered somewhat by the lack of water solubility. In the absence of hydrolysis, sorption to solids (e.g., sludge and sediments) will be the primary mechanism of removal. Hydrolysis products are predominantly insoluble stable polyureas. Biodegradation is minimal for most compounds and volatilisation is negligible.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
	No Data available for all ingredients	No Data available for all ingredients

Bioaccumulative potential

Ingredient	Bioaccumulation		
	No Data available for all ingredients		
Mobility in soil			
Mobility in soil Ingredient	Mobility		

SECTION 13 Disposal considerations

Waste treatment methods	
Product / Packaging disposal	 Containers may still present a chemical hazard/ danger when empty. Return to supplier for reuse/ recycling if possible. Otherwise: If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill. Where possible retain label warnings and SDS and observe all notices pertaining to the product. Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked. A Hierarchy of Controls seems to be common - the user should investigate: Reduction Reuse Recycling Disposal (if all else fails) This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. DO NOT allow wash water from cleaning or process equipment to enter drains. It may be necessary to collect all wash water for treatment before disposal. In all cases disposal to eswer may be subject to local laws and regulations and these should be considered first. Where in doubt contact the responsible authority. DO NOT recycle spilled material. Consult State Land Waste Management Authority for disposal. Neutralise spill material carefully and decontaminate empty containers and spill residues with 10% ammonia solution plus detergent or a proprietary decontaminate prior to disposal. DO NOT seal or stopper drums being decontaminated as CO2 gas is generated and may pressurise containers.

Ensure that the hazardous substance is disposed in accordance with the Hazardous Substances (Disposal) Notice 2017

Disposal Requirements

Packages that have been in direct contact with the hazardous substance must be only disposed if the hazardous substance was appropriately removed and cleaned out from the package. The package must be disposed according to the manufacturer's directions taking into account the material it is made of. Packages which hazardous content have been appropriately treated and removed may be recycled.

The hazardous substance must only be disposed if it has been treated by a method that changed the characteristics or composition of the substance and it is no longer hazardous.

SECTION 14 Transport information

Labels Required Marine Pollutant NO HAZCHEM Not Applicable

Land transport (UN): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

This substance is to be managed using the conditions specified in an applicable Group Standard

HSR Number	Group Standard
HSR002675	Surface Coatings and Colourants (Toxic [6.1]) Group Standard 2017
_	

of Chemicals - Classification Data

New Zealand Inventory of Chemicals (NZIoC)

New Zealand Workplace Exposure Standards (WES)

New Zealand Hazardous Substances and New Organisms (HSNO) Act - Classification

polymeric diphenylmethane diisocyanate is found on the following regulatory lists

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

New Zealand Approved Hazardous Substances with controls

New Zealand Hazardous Substances and New Organisms (HSNO) Act - Classification of Chemicals

Hazardous Substance Location

Subject to the Health and Safety at Work (Hazardous Substances) Regulations 2017.

Hazard Class	Quantity (Closed Containers)	Quantity (Open Containers)
Not Applicable	Not Applicable	Not Applicable

Certified Handler

Subject to Part 4 of the Health and Safety at Work (Hazardous Substances) Regulations 2017.

Class of substance	Quantities
6.1A, 6.1B, 6.1C (except for propellant powders of classes 1.1C (UN 0160) and 1.3C (UN 0161)	Any quantity

Refer Group Standards for further information

Tracking Requirements

Subject to tracking according to the Health and Safety at Work (Hazardous Substances) Regulations 2017

- Refer to the regulation for more information

National Inventory Status

National Inventory	Status	
Australia - AIIC	Yes	
Australia Non-Industrial Use	No (polymeric diphenylmethane diisocyanate)	
Canada - DSL	Yes	
Canada - NDSL	No (polymeric diphenylmethane diisocyanate)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	No (polymeric diphenylmethane diisocyanate)	
Japan - ENCS	Yes	
Korea - KECI	Yes	
New Zealand - NZIoC	Yes	
Philippines - PICCS	Yes	
USA - TSCA	Yes	
Taiwan - TCSI	Yes	

RS- 3m Scotchcast 470W Resin (Part B) #811-2760, #811-2772, #811-2776, #811-2782, #811-2785, #811-2788 (NZ)

National Inventory	Status
Mexico - INSQ	Yes
Vietnam - NCI	Yes
Russia - ARIPS	Yes
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

SECTION 16 Other information

Revision Date	15/07/2020
Initial Date	15/07/2020

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

PC – TWA: Permissible Concentration-Time Weighted Average PC – STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit. IDLH: Immediately Dangerous to Life or Health Concentrations OSF: Odour Safety Factor NOAEL : No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

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